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Fully Solution-Processed Low-Voltage Aqueous In₂O₃ Thin-Film Transistors Using an Ultrathin ZrO_x Dielectric

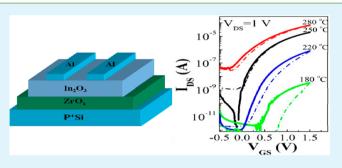
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Supporting Information

ABSTRACT: We reported here "aqueous-route" fabrication of In_2O_3 thin-film transistors (TFTs) using an ultrathin solution-processed ZrO_x dielectric thin film. The formation and properties of In_2O_3 thin films under various annealing temperatures were intensively examined by thermogravimetric analysis, Fourier transform infrared spectroscopy, and atomic force microscopy. The solution-processed ZrO_x thin film followed by sequential UV/ozone treatment and low-temperature thermal-annealing processes showed an amorphous structure, a low leakage-current density (~1 × 10⁻⁹ A/cm² at 2 MV/cm), and a high breakdown electric field (~7.2 MV/



cm). On the basis of its implementation as the gate insulator, the In_2O_3 TFTs based on ZrO_x annealed at 250 °C exhibit an on/ off current ratio larger than 10⁷, a field-effect mobility of 23.6 cm²/V·s, a subthreshold swing of 90 mV/decade, a threshold voltage of 0.13 V, and high stability. These promising properties were obtained at a low operating voltage of 1.5 V. These results suggest that "aqueous-route" In_2O_3 TFTs based on a solution-processed ZrO_x dielectric could potentially be used for low-cost, low-temperature-processing, high-performance, and flexible devices.

KEYWORDS: aqueous solution process, low-temperature process, ultrathin zirconium oxide, indium oxide, thin-film transistor

morphous metal oxide semiconductors have been Aextensively studied as the channel materials for thin-film transistors (TFTs) in display backplanes and other optoelectronic devices.^{1,2} Recently reported electrical parameters of metal oxide TFTs, especially the carrier mobility, are generally superior to those of amorphous silicon-based TFTs and, in some cases, even comparable to those of polycrystalline siliconbased TFTs.^{3,4} However, these TFTs were typically manufactured by vacuum-based methods. Although the vacuum-based deposition methods have their own advantages, the high fabrication cost and large-area device uniformity restrict the areas of their applications.⁵ For this consideration, considerable researches have been conducted on the development of a solution process for the construction of metal oxide TFTs.⁵⁻⁷ It is noted that, in most of the previous reports of solutionprocessed TFTs, a high-temperature annealing process is imperative to achieving reasonable electrical properties. To decrease the annealing temperature of solution-processed oxide layers, several research groups have proposed novel approaches to achieve oxide TFTs at temperatures under 300 °C, including sol-gel on chip,⁸ a chemical energetic combustion process through an oxidizer and fuel,9 a UV/ozone photoannealing method,¹⁰ annealing in an O_2/O_3 atmospheric environment,¹ and zinc hydroxoamine complex precursors.¹² However, most of these approaches fabricated the TFTs on thermally grown or vacuum-deposited SiO_2 dielectrics, which will certainly result in high operating voltages (larger than 40 V).

It is known that the use of appropriate solvents, metal precursors, and gate dielectrics is crucial to enable low processing temperature and high device performances. Park et al. proposed an "aqueous route" to fabricate metal oxide TFTs with the maximum processing temperature not exceeding 350 °C.6 Because water is used as a solvent, instead of the often-used 2-methoxyethanol (2-ME)-based organic solvents, the "aqueous-route" synthesis is considered to be healthier, safer, and environmentally friendlier. Unlike several conventionally used 2-ME-based precursors, the aqueous solutions are insensitive to ambient moisture. The inert atmosphere to store and handle a precursor solution is not necessary.¹³ More importantly, because the coordinating bond between the metal cation and neighboring aquo ion is relatively weak (electrostatic interaction), it is easily broken with low thermal energy compared with the covalent bonds in the 2-ME-based precursor.⁵ Therefore, "aqueous route" is considered to be a promising technique to fabricate the metal oxide layers at low temperature.

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As a candidate of channel material, indium oxide (In_2O_3) has been widely studied because it can provide high electron mobility originating from the ns orbital of indium. The ns orbital is larger than the 2p orbital of the oxygen anion.^{1,14} Meanwhile, In_2O_3 is one of the semiconductor materials that can be achieved by the solution process and exhibits various electrical performances depending on the stoichiometry and defects in materials.¹¹ When the advantages of "aqueous route" and In_2O_3 materials are combined, aqueous In_2O_3 is considered to be an ideal candidate to fabricate the high-mobility channel layer for TFT devices at low temperature.

Moreover, the gate dielectric plays an equally critical role in determining the electrical performances of the oxide TFTs. Generally, gate dielectrics are required to be smooth, dense, and pinhole-free to allow a low leakage current and a high breakdown electric field.⁷ However, for metal oxide dielectrics fabricated at low temperature (<300 °C), it is difficult to achieve the aforementioned purposes.^{14,15} In our previous report, a UV/ozone-treated ZrO_x dielectric thin film was successfully fabricated to replace SiO₂ as the dielectric for InTiZnO TFTs.¹⁶ The smooth surface and excellent electrical performances of the ZrO_x dielectric guaranteed that the asfabricated TFTs exhibited decent characteristics at a low operating voltage of 3 V.

On the basis of our previous solution process for oxide dielectrics, we further demonstrate the fabrication of high-performance and low-operating-voltage TFTs using aqueous In_2O_3 as channel layers. The effects of the annealing temperature on the thermal behavior, chemical properties, surface morphologies, and performances of the TFT devices were also studied.

The detailed experimental section and instrumental analysis are shown in sections S1 and S2 in the Supporting Information (SI). To understand the thermal behaviors of ZrO_x and In_2O_3 precursor solutions, thermogravimetric analysis (TGA) measurements were performed, and the results are shown in Figure 1. In general, the hydrolysis reaction of precursor xerogel occurred in the temperature range of 100-150 °C.⁶ In this experiment, because the indium precursor was dissolved in water, the ionized indium cation was solvated by the neighboring water molecules. Because the coordination number of In^{3+} was 6, it was considered to be a true coordination

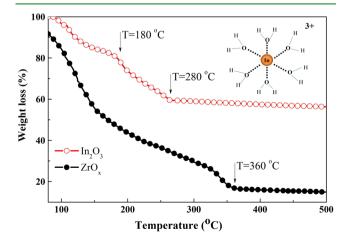


Figure 1. Thermal behaviors of In_2O_3 and ZrO_x dried precursor solutions. The inset represents the proposed indium complex in aqueous solution.

complex, with six water molecules acting as σ -donating ligands.¹³ For In₂O₃ xerogel, dehydroxylation of indium hydroxide began at ~180 °C and was nearly complete when the xerogel was annealed at temperatures higher than 280 °C. On the basis of this dehydroxylation behavior, the annealing condition for aqueous In₂O₃ thin films was estimated to be from 180 to 280 °C. For ZrO_x xerogel, no significant weight loss was observed at temperatures above 360 °C. In the previously reported literature, UV/ozone treatment has been proven to be effective in deceasing the decomposition temperature for metal oxide thin films.^{10,17} However, there have been no reports applying this technique to fabricate the dielectrics. In this work, a ZrO_r dielectric was processed using a UV/ozone pretreatment before the thermal-annealing process. Meanwhile, in order to reduce the thermal budget (indium diffusion and miscibility phenomenon) brought by the annealing treatment for In2O3 channel layers and to guarantee a low-temperature process, we conducted the annealing process for a ZrO_x dielectric at 300 °C.

To better understand the formation of In_2O_3 thin films with aforementioned annealing conditions, Fourier transform infrared (FT-IR) measurements were carried out, and the results are shown in Figure 2. Several vibration peaks are observed in the

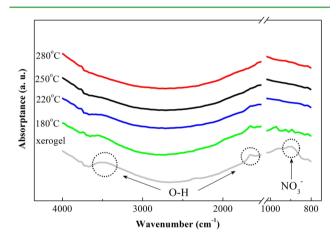


Figure 2. FT-IR spectra of the In_2O_3 thin films annealed at different temperatures.

In₂O₃ xerogel film. The broad peaks in the ranges of 3300– 3500 and 1500–1700 cm⁻¹ indicate O–H stretching vibrations.⁶ The peak at 850 cm⁻¹ is related to the existence of NO₃^{-.18} As the annealing temperature was increased to 180 °C, the nitrate groups were pyrolyzed, while a large amount of hydroxyl species still remained in the thin films. In the case of annealing temperatures higher than 250 °C, the O–H vibration peaks disappeared, and the spectrum was similar to that of the substrate. This indicated that the metal hydroxides [In- $(OH_2)_6^{3+}$] were converted to metal oxides (In₂O₃). In view of the TGA and FT-IR results, it was proven that the weight loss in the temperature range of 80–180 °C mainly originated from the decomposition of nitrate groups and the continuous decrease was caused by the dehydroxylation reaction.

Parts a and b of Figure 3 show the atomic force microscopy (AFM) image of a ZrO_x thin film and the root-mean-square (RMS) values of In_2O_3 thin films on ZrO_x annealed at different temperatures, respectively. The X-ray diffraction (XRD) patterns of various In_2O_3 thin films are shown in Figure S1 in the SI. The ZrO_x thin film exhibits a smooth surface with a

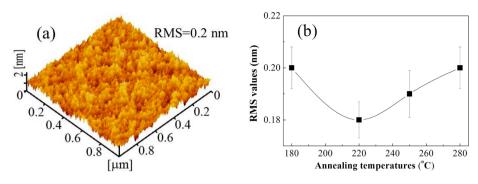


Figure 3. (a) AFM image of a ZrO_x thin film. (b) RMS values of In₂O₃ thin films on ZrO_x annealed at different annealing temperatures.

roughness of 0.2 nm and free pinholes. Prior to the thermalannealing process, the UV/ozone treatment allowed the volatile gases, originally from solvent, to overflow slowly through the surface of the gel thin films. Therefore, the surface of the ZrO_x thin film with a UV/ozone pretreatment was kept as smooth as possible. A smooth surface is beneficial for charge-carrier transportation in semiconductors because a rough interface could induce physical traps or disturb the growth of channel layers. The average RMS values of In_2O_3 thin films on ZrO_x annealed at various temperatures were measured as 0.20, 0.18, 0.19, and 0.20 nm, respectively. These small surface roughness values are beneficial from the smooth surface of the ZrO_x thin film and are ideal for obtaining a high device performance.

To characterize the dielectric and electrical properties of the ZrO_x thin films, a capacitor with the structure of $Al/ZrO_x/p^+$ -Si was employed. Figure 4 shows the areal capacitance of the ZrO_x

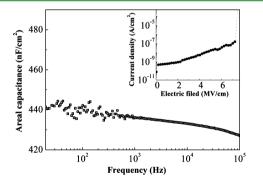


Figure 4. Areal capacitance and dielectric constant of an $Al/ZrO_x/p^{+2}$ silicon capacitor as a function of the frequency. The leakage-current density versus electric field is shown in the inset.

capacitor in the frequency range between 20 Hz (440 nF/cm²) and 100 kHz (426 nF/cm²). Such a large capacitance density can effectively decrease the operating and subthreshold voltages of the TFT devices. In addition, the ZrO_x dielectric showed a small frequency dispersion of the capacitance (~3.2%), indicating a low defect density such as hydroxyl group and/or oxygen vacancies in the thin film.¹⁹ This is undoubtedly beneficial to the leakage current because the conduction paths in dielectrics are mainly caused by hydroxyl and grain boundaries. During the fabrication of high-k dielectrics on a silicon wafer using chemical methods, the presence of SiO₂ interface layer (IL) can cause a decrease of the k value. In order to calculate the dielectric constant of the ZrO_x thin film, the influence of the SiO_2 IL between ZrO_x and the silicon wafer must be excluded. Assuming that the IL is SiO₂, the effective dielectric constant, estimated using a series capacitor model (1/

 C_{SiO_2} + $1/C_{ZrO_2}$ = $1/C_{total}$), was calculated to be around 12.5. The inset of Figure 4 presents the leakage-current density (I_{leak}) of the ZrO_x capacitor at various electric fields. The capacitor exhibits a low J_{leak} of 1×10^{-9} A/cm² at 2.0 MV/cm and a high dielectric breakdown electric field of 7.2 MV/cm. Jleak is nearly 2 orders of magnitude smaller than the previously published results of the sol-gel-derived ZrO_x dielectrics that were annealed at 300 $^\circ C$ or were treated by UV/ozone. 20,21 The low leakage-current density and high breakdown electric field are due to not only the smooth surface, dense structure, and high oxidation states of the ZrO_x thin film but also its amorphous structure (shown in Figure S2 in the SI). All of the properties suggest that the UV/ozone treatment followed by a low-temperature annealing process have great potential for the gate dielectrics in fabricating the low-voltage, high-performance oxide TFTs.

To investigate the performance of aqueous In₂O₃ TFTs integrated on solution-processed high- $k \operatorname{ZrO}_{x}$ dielectric, TFT devices with bottom-gate and top-contact architecture were fabricated. The output curves are shown in Figure S3 in the SI, and the summarized output curves of the In_2O_3 TFTs at a V_{GS} of 1.5 V are depicted in Figure 5a. These devices exhibit typical n-channel behavior with clear pinch-off voltage and current saturation. It can be seen that the operating voltage is only 1.5 V, which is important for low-power electronics. Figure 5b shows the corresponding transfer characteristics of In₂O₃ TFTs with a double-sweep gate voltage model. All of these TFTs show hysteresis characteristics, and their direction is clockwise. When the annealing temperature was 180 °C, the device exhibited poor performance, including a low saturation current of 3 \times 10 $^{-9}$ A and a large hysteresis behavior with a 0.34 V voltage shift. The large hysteresis window and low saturation currents are mainly due to the number of defect states and the degree of oxidation in the semiconductor channel layers.² According to the results of FT-IR, the incomplete decomposition of nitrate groups and large amounts of hydroxide undoubtedly degraded the performance of the TFT device. With the accelerated dehydroxylation reaction and the formation of In-O bonds in In₂O₃ channel layers at higher annealing temperatures, the saturation current increased from 8.5×10^{-7} A (220 °C) to 8.7×10^{-5} A (280 °C) and hysteresis windows became negligible.

To further investigate the electrical properties of the asfabricated In₂O₃ TFTs, the threshold voltage ($V_{\rm TH}$) was defined by fitting a straight line to the plot of the square root of $I_{\rm DS}$ versus $V_{\rm GS}$. The field-effect mobility ($\mu_{\rm FE}$) can be calculated in the saturation region ($V_{\rm DS} > V_{\rm G} - V_{\rm TH}$) using the following equation:

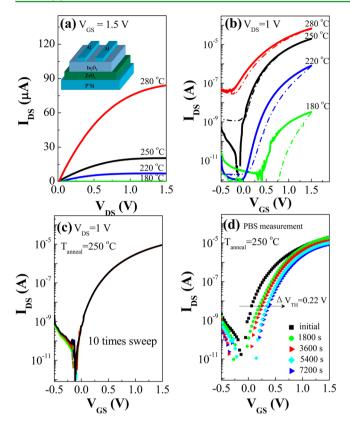


Figure 5. (a) Summarized output curves of In_2O_3 TFTs on ZrO_x annealed at various annealing temperatures. (b) Corresponding transfer characteristics of In_2O_3 TFTs at $V_D = 1$ V as a function of the annealing temperature. (c) 10 times sweep test for a 250 °C annealed In_2O_3 TFT. (d) Transfer curves of a 250 °C annealed In_2O_3 TFT under PBS with a V_{GS} value of 1.5 V for 7200 s.

$$I_{\rm DS} = \frac{1}{2} \frac{W}{L} C_{\rm i} \mu_{\rm FE} (V_{\rm GS} - V_{\rm TH})^2$$
(1)

where C_{ν} L, and W are the areal capacitance of the ZrO_x dielectric and the channel length and width of the TFT, respectively. The electrical parameters of the TFTs are summarized in Table I. The $\mu_{\rm FE}$ values of the In₂O₃ TFTs annealed at 220, 250, and 280 °C were calculated to be 1.1, 23.6, and 29.5 cm²/V·s, respectively. It is known that the conduction band minimum (CBM) in metal oxide semiconductors should be primarily composed of dispersed vacant s states with short interaction distances for efficient carrier transportation, which can be achieved in ionic oxide but not obviously in hydroxide.²³ Therefore, the dehydroxylation reaction at higher annealing temperatures for In₂O₃ channel layers not only increased the saturation current but also enhanced the $\mu_{\rm FE}$ value of the as-fabricated TFTs.

It can be clearly seen from Table I that the electrical performance of the In_2O_3 TFTs annealed at 250 °C showed the best performance, including a reasonable μ_{FE} of 23.6 cm²/V·s, a

high on/off current ratio $(I_{\rm on}/I_{\rm off})$ of 1.1×10^7 , a low $V_{\rm TH}$ of 0.13 V, and a small subthreshold swing (SS) of 90 mV/dec Generally, the SS values directly reflect the switching speed and power consumption of the TFT devices. The small SS values for all of these TFTs were beneficial from the large areal capacitance of the ZrO_x dielectric and the electronic-clean interface between In_2O_3 and ZrO_x . Although a high saturation current and large $\mu_{\rm FE}$ were obtained in the case of the In₂O₃ TFT annealed at 280 °C, it was found to operate in the depletion mode with a negative $V_{\rm TH}$ of -0.27 V because of high carrier concentration arising from the Fermi level proximity to the CBM.2 The high carrier concentration, together with enhancement of the interface defects in the dielectric channel interface, will make it difficult to deplete the In2O3 channel layer, leading to a negative $V_{\rm TH}$, a high off current, and a low $I_{\rm op}/I_{\rm off}$ value. On the basis of the results of the In₂O₃ TFTs annealed at various temperatures, it can be deduced that appropriate self-doping (especially hydroxyl species) in the In₂O₃ system at low temperature is needed to control the excess carrier concentration and to optimize the dielectric/ channel interface and so the electrical performances of the TFT devices. To our best knowledge, this is the lowest reported operating voltage for fully solution-processed TFTs. In this study, the aqueous route allows fabrication of the In₂O₃ TFTs at much lower temperature compared with the previously reported 2-ME-based approach.^{11,24} This result represents a significant step toward the development of nontoxic, low-cost, and large-area oxide electronics.

The degradation behavior of the In₂O₃ TFT annealed at 250 $^{\circ}\mathrm{C}$ was also investigated by measuring the V_{TH} shift in consecutive transfer curves. This simple procedure is useful to analyze the early-stage aging of devices and also to deduce the instability mechanism that might be present.²⁵ Figure 5c shows the results of a consecutive 10-time sweep test at $V_{\rm D} = 1$ V. High stability without any degradation was exhibited in this test. This result can be attributed to two reasons: one is due to the fact that a low density of trap states exists at the $In_2O_3/$ ZrO_{*} interface (as indicated in the AFM results); the other is due to the fact that the background carrier concentration in the In₂O₃ channel layer is enough to fill the traps at zero-gate bias. Meanwhile, a positive bias stress (PBS) test for a 250 °C annealed In2O3/ZrOx TFT was performed. The device was stressed under the following conditions: V_{GS} and V_{DS} were set to 1.5 and 1.0 V, respectively; the stress duration was 7200 s. The threshold voltage shift ($\Delta V_{\rm TH}$) in the transfer curves under the PBS test is shown in Figure 5d. The parallel V_{TH} shift of the transfer curve (~ 0.22 V), with a negligible change in the SS value is considered to be from the carrier-trapping mechanism.²⁶ The hysteresis measurement and consequence sweep test in transfer curves indicate very few trap defects existing in the 280 °C annealed In₂O₃ channel layer and In₂O₃/ZrO_x interface. Therefore, the charge-trapping model alone cannot entirely account for the results obtained. Jeong et al.²⁷ and Pan et al.²⁸ reported that the interaction between the channel layer

Table I. Electrical Parameters of In₂O₃ TFTs on a ZrO_x Dielectric Annealed at Various Temperatures

In_2O_3 annealing temperature (°C)	$\mu_{\rm FE}~({ m cm}^2/{ m V}{ m \cdot}{ m s})$	$I_{\rm on}/I_{\rm off}$	V_{TH} (V)	SS (mV/dec)	hysteresis (V)
180		$10^3 - 10^4$	0.71 ± 0.05	160 ± 15	0.41 ± 0.09
220	1.1 ± 0.4	$10^{5} - 10^{6}$	0.58 ± 0.03	110 ± 10	0.33 ± 0.05
250	23.6 ± 0.3	$\sim 10^{7}$	0.13 ± 0.02	90 ± 10	0.05 ± 0.02
280	29.5 ± 0.9	$10^3 - 10^4$	-0.27 ± 0.06	390 ± 20	0.09 ± 0.03

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and oxygen in an ambient atmosphere plays a critical role in determining the $V_{\rm TH}$ instability. When the PBS test was applied in the atmosphere, excess electrons accumulated in the channel layer. The surrounding oxygen molecules have large electron negativity, which can capture electrons from the conduction band to form O^{2-} species. The adsorption of oxygen molecules in the channel layer can deplete the electron carriers, leading to a positive shift of $\Delta V_{\rm TH}^{29}$

In summary, we have integrated an aqueous route with solution deposition of oxide-based TFTs. An ultrathin solutionprocessed ZrO, thin film was fabricated as the dielectric, which was processed by a UV/ozone treatment and a low-temperature annealing process. Such a ZrO_x layer exhibited excellent electrical performance as a gate insulator, such as a high dielectric constant of ~12.5, a low leakage-current density of 1 \times 10⁻⁹ A/cm² at 2.0 MV/cm, and a high breakdown electric field of 7.2 MV/cm. The optimized In₂O₃ TFT, which was annealed at 250 °C, exhibited high performances with a $\mu_{\rm FE}$ of 23.6 cm²/V·s, an $I_{\rm on}/I_{\rm off}$ value of 1.1×10^7 , $V_{\rm TH}$ of 0.13 V, and SS of 90 mV/decade. All of these decent electrical performances were obtained at a low operating voltage of 1.5 V. The "aqueous route" is applicable to a broad range of amorphous metal oxide compositions, and it should also provide a new approach for integrating more amorphous oxide materials into functional electronic and optoelectronic devices.

ASSOCIATED CONTENT

S Supporting Information

Experimental section, instrumental analysis, XRD patterns of water-induced In_2O_3 thin films annealed at various annealing temperatures and of a ZrO_x thin film, output curves of In_2O_3/ZrO_x TFTs as a function of the annealing temperature, and leakage-current density versus electric field curves of ZrO_x thin films annealed at various conditions. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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